## In Vivo Metabolism of 15α-Hydroxyprogesterone in Man\*

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ABSTRACT: Labeled  $15\alpha$ -hydroxyprogesterone was prepared by the microbiological hydroxylation of [7-3H]progesterone or [4-14C]progesterone. Nonlabeled  $15\alpha$ -hydroxyprogesterone was administered orally to normal male subjects with or without the simultaneous intravenous injection of the labeled steroid, and the urinary conjugates were hydrolyzed with Glusulase. The following six urinary metabolites were isolated and identified in these studies:  $15\alpha$ -hydroxyprogesterone,  $3\alpha$ ,  $15\alpha$ -dihydroxy- $5\alpha$ -pregnan-20-one,  $3\alpha$ ,  $15\alpha$ -dihydroxy- $5\beta$ pregnan-20-one,  $3\beta$ ,  $15\alpha$ -dihydroxy- $5\beta$ -pregnan-20-one,  $5\alpha$ pregnane- $3\alpha$ ,  $15\beta$ ,  $20\beta$ -triol, and  $5\beta$ -pregnane- $3\alpha$ ,  $15\alpha$ ,  $20\beta$ -triol. When a trace amount of  $[4-14C]15\alpha$ -hydroxyprogesterone was administered to a subject in the third trimester of pregnancy and to a nonpregnant subject in the luteal phase of the menstrual cycle,  $3\beta$ ,  $15\alpha$ -dihydroxy- $5\beta$ -pregnan-20-one was isolated from the sulfate fraction and  $15\alpha$ -hydroxyprogesterone,  $3\alpha,15\alpha$ -dihydroxy- $5\alpha$ -pregnan-20-one,  $3\alpha,15\alpha$ -dihydroxy- $5\beta$ - pregnan-20-one, and  $5\alpha$ -pregnane- $3\alpha$ ,  $15\alpha$ ,  $20\beta$ -triol were isolated from the glucosiduronate fraction of the urine of both subjects.

The specific activities of the metabolites isolated from the urine of the pregnant subject were from 25 to 45 times lower than the specific activity of the injected  $15\alpha$ -hydroxyprogesterone but the same metabolites isolated from the urine of the nonpregnant subject had specific activities which were the same as that of the injected steroid. These results demonstrate that  $15\alpha$ -hydroxyprogesterone and its metabolites are normally excreted in the urine of the pregnant subject but not in the urine of the nonpregnant subject. The studies reported also demonstrate that the  $15\alpha$ -hydroxyl group partially inhibits the reduction of the  $\Delta^4$ -3-ketone and the C-20 ketone, leads to an increased ratio of  $5\alpha$ : $5\beta$ -reduced urinary metabolites, and directs the reduction of the C-20 ketone toward the  $20\beta$ -alcohol.

In a previous communication (Giannopoulos and Solomon, 1967) it was shown that  $15\alpha$ -hydroxyprogesterone is a constituent of human late-pregnancy urine. Because the  $16\alpha$ -hydroxyl group on the steroid nucleus alters the *in vivo* metabolism of C-19 steroids (YoungLai and Solomon, 1967) and modifies to a lesser extent the metabolism of steroids in the C-21 series (Fukushima *et al.*, 1961, 1967; Hirschmann *et al.*, 1961; Calvin and Lieberman, 1962; Ruse and Solomon, 1966; Romanoff *et al.*, 1967), it was therefore of some interest to investigate the metabolic fate of  $15\alpha$ -hydroxyprogesterone administered to humans. We chose to study the metabolism of  $15\alpha$ -hydroxyprogesterone in the pregnant subject during the third trimester because of the findings that the largest amounts of this steroid are excreted in the urine during late

pregnancy (Giannopoulos et al., 1969). As a comparative study, we also investigated the in vivo metabolism of  $15\alpha$ -hydroxyprogesterone in a nonpregnant subject during the luteal phase of the menstrual cycle.

In the initial stages of the studies to be reported it was necessary to obtain sufficient amounts of the urinary metabolites of  $15\alpha$ -hydroxyprogesterone for their identification and to serve as carrier steroids in subsequent work, because such metabolites had not hitherto been described. In order to accomplish this, two studies were performed in which large amounts of nonlabeled steroid were fed by mouth to normal male volunteers and the urinary metabolites were isolated and identified. In one of these studies the labeled steroid was also injected intravenously.

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the Medical Research Council of Canada (MT-1658).

### Materials and Methods

Only those procedures which have not been previously described (Ruse and Solomon, 1966; YoungLai and Solomon, 1967) will be given.

Chromatography. Solvent systems used for chromatography are shown in Table I. All the chromatograms were developed at room temperature.

Standardization of Labeled Acetic Anhydride. One solution of [1-14C]acetic anhydride and two solutions of [3H]acetic anhydride, 20% (v/v) in benzene, were standardized by the acetylation of deoxycorticosterone, chromatography of the acetate on an alumina column, and crystallization to constant specific activity. The [1-14C]acetic anhydride and solutions 1 and 2 of [3H]acetic anhydride had specific activities of 2.30  $\times$  104, 3.26  $\times$  107, and 3.68  $\times$  108 dpm per mg of deoxycorticosterone acetate, respectively.

Formation of Derivatives. A derivative of  $15\alpha$ -acetoxypro-

¹ The following trivial names and abbreviations are used: progesterone, pregn-4-ene-3,20-dione;  $20\alpha$ -dihydroprogesterone,  $20\alpha$ -hydroxypregn-4-en-3-one;  $20\alpha$ -dihydroprogesterone acetate,  $20\alpha$ -acetoxypregn-4-en-3-one;  $20\beta$ -dihydroprogesterone,  $20\beta$ -dihydroprogesterone acetate,  $20\beta$ -acetoxypregn-4-en-3-one;  $15\alpha$ -hydroxyprogesterone,  $15\alpha$ -hydroxypregn-4-ene-3-20-dione;  $15\alpha$ -acetoxyprogesterone,  $15\alpha$ -acetoxypregn-4-ene-3,20-dione;  $15\alpha$ -dihydroxypregn-4-en-3-one;  $15\alpha$ -hydroxyprogesterone,  $15\alpha$ -20 $\beta$ -diacetoxyprogesterone,  $15\alpha$ -dihydroxypregn-4-en-3-one;  $15\alpha$ -hydroxypregn-enolone,  $3\beta$ ,  $15\alpha$ -dihydroxypregn-5-en-20-one;  $16\alpha$ -hydroxypregn-4-ene-3,20-dione;  $15\alpha$ -hydroxypregn-4-ene-3,20-dione;  $15\alpha$ -hydroxydeoxycorticosterone,  $15\alpha$ ,21-dihydroxypregn-4-ene-3,20-dione;  $\beta$ -DPNH, dihydro- $\beta$ -diphosphopyridine nucleotide disodium salt; DDQ, 2,3-dichloro-5,6-dicyanobenzoquinone.

TABLE 1: Solvent Systems Used in Chromatography.

System	Type of Chromatography	Solvent Mixture
		Talana mandan alual
Α	Ppc⁴	Toluene-propylene glycol
В	Ppc	Skellysolve B-methanol-water (10:9:1)
С	Ppc	2,2,4-Trimethylpentane-toluene- methanol-water (5:5:7:3)
D	Cccb	n-Heptane-ethyl acetate-methanol- water (10:10:13:7)
E	Ccc	n-Hexane-ethyl acetate-methanol-water (10:10:13:7)
F	Ppc	Benzene-methanol-water (20:11:9)
G	Ppc	Toluene-ethyl acetate-methanol-water (9:1:6:4)
H	Tlc∘	Chloroform-ethanol (9:1)
J	Tlc	Benzene-ethanol (1:1)
K	Tlc	Benzene-ethyl acetate (2:1)

<sup>&</sup>lt;sup>a</sup> Ppc, paper partition chromatography. <sup>b</sup> Ccc, Celite column chromatography. <sup>c</sup> Tlc, thin-layer chromatography.

gesterone was prepared by reduction of the steroid with NaBH<sub>4</sub> as described by Norymberski and Woods (1955) and subsequent oxidation of the allylic alcohol at C-3 with DDQ as described by Burn et al. (1960). The infrared spectrum (CS<sub>2</sub>) of the crystalline product obtained indicated a retention of the acetate group (1738 cm<sup>-1</sup>), the presence of the  $\Delta^4$ -3ketone (1680 cm<sup>-1</sup>), and of a hydroxyl group (3600 cm<sup>-1</sup>), and the loss of the 20-ketone. This product had a melting point of 192-193° and absorbed ultraviolet light. The infrared spectrum of this derivative was identical with the spectrum of the product obtained after reduction of  $15\alpha$ -acetoxyprogesterone with 20β-hydroxysteroid dehydrogenase as described below. In addition, optical rotation studies, described in the Experimental Section, demonstrated that NaBH<sub>4</sub> reduction of  $15\alpha$ -acetoxyprogesterone favors the formation of the  $20\beta$ alcohol. These results permit the assignment of the structure  $15\alpha$ -acetoxy-20 $\beta$ -hydroxypregn-4-en-3-one to the product obtained by NaBH4 reduction and subsequent DDO oxidation of  $15\alpha$ -acetoxyprogesterone.

Reduction of  $15\alpha$ -acetoxyprogesterone with  $20\beta$ -hydroxysteroid dehydrogenase was performed as follows. An aliquot of  $15\alpha$ -acetoxyprogesterone (3.4 mg) was dissolved in 0.4 ml of ethanol and the solution was added slowly to 20 ml of 0.1 м Tris-HCl and 0.0027 м EDTA buffer (pH 7.6). Following the addition of 194 mg of  $\beta$ -DPNH (Sigma) and 0.2 ml (1.0 mg) of 20β-hydroxysteroid dehydrogenase, Type 11 (Sigma, lot 97B-0490, prepared from Streptomyces hydrogenans, and stored at 0-5° as a suspension in 2.2 M ammonium sulfate, 5 mg/ml), the mixture was incubated at 25° in a stoppered flask for 2 hr. The reaction was stopped by the addition of 50 ml of distilled water and the steroids were extracted with ethyl acetate to give a residue weighing 2.8 mg. It was chromatographed on thin-layer chromatography using system K and two ultraviolet-positive bands were observed at average distances of 4.8 and 6.3 cm. The more polar band was eluted

TABLE II: Proof of Radiochemical Purity of  $[4-14C]15\alpha$ -Hydroxyprogesterone Prepared Microbiologically.

	Specific Activities (dpm/mg)			
Crystn	Crystals	MLª		
1	2250	2220		
2	2180	2140		
3	2200	2240		
Calcd <sup>b</sup>	2330			

<sup>a</sup> ML, mother liquors. <sup>b</sup> An aliquot of the material eluted from paper containing  $1.12 \times 10^5$  dpm was mixed with 48.0 mg of carrier  $15\alpha$ -hydroxyprogesterone prior to crystallization.

to give a residue (2.3 mg) which was crystallized from acetone–Skellysolve B to yield 1.5 mg of coarse needles, mp 192–193°. Its infrared spectrum (CS<sub>2</sub>) indicated the retention of the acetate group, the presence of the  $\Delta^4$ -3-ketone, the loss of the 20-ketone, and the appearance of a hydroxyl band at 3590 cm<sup>-1</sup>. This spectrum was identical with the spectrum of the material obtained following reduction of  $15\alpha$ -acetoxyprogesterone with NaBH<sub>4</sub> and oxidation of the resulting product with the DDQ reagent as described above. The material obtained after incubation of  $15\alpha$ -acetoxyprogesterone with  $20\beta$ -hydroxysteroid dehydrogenase was therefore assigned the structure of  $15\alpha$ -acetoxy- $20\beta$ -hydroxypregn-4-en-3-one.

Nuclear Magnetic Resonance Spectroscopy. The nuclear magnetic resonance spectra were run on a Varian HR-100 spectrometer using deuteriochloroform (CDCl<sub>3</sub>) as the solvent and tetramethylsilane as an internal standard. A glass microcell (NMR Specialties) of approximately 50-μl volume was used in these determinations. Chemical shift measurements were obtained from individual side-band measurements, using tetramethylsilane as the reference standard. In some instances a Varian HR-100 spectrometer was used, and since the instrument was field-frequency locked, the spectra were displayed on precalibrated charts. Observed chemical shifts were compared with shifts calculated for likely structures from the additive substituent effects given in Zürcher's Tables (Zürcher, 1963).

Mass Spectra. Mass spectra were run by Dr. Alan Duffield, Department of Chemistry, Stanford University. These spectra were determined using an Atlas CH-4 mass spectrometer run at 70 eV using a direct inlet system with the ion source at 180°.

Hydrolysis of Urinary Conjugates. Steroid conjugates were hydrolyzed by the modification of the solvolytic procedure of Jacobsohn and Lieberman (1962) and then with  $\beta$ -glucuronidase as previously described (YoungLai and Solomon, 1967). In some instances steroid conjugates were hydrolyzed by Glusulase as described by Ruse and Solomon (1966).

 $[7^{-3}H]15\alpha$ -Hydroxyprogesterone. The preparation, purification, radiochemical purity, and the determination of the specific activity of  $[7^{-3}H]15\alpha$ -hydroxyprogesterone have been described (Giannopoulos and Solomon, 1967). The specific activity of this material was  $1.74 \times 10^7$  dpm/mg.

 $[4-^{14}C]15\alpha$ -Hydroxyprogesterone. The microorganism used in this preparation was a strain of penicillium (ATCC 11598)

TABLE III: Determination of the Specific Activity of  $[4^{-1}C]15\alpha$ -Hydroxyprogesterone.

Crystn	Specific Activities (dpm of <sup>8</sup> H/mg)									
	15α-Acetoxyprogesterone <sup>a</sup>				15α-Acetoxy-20β-hydroxypregn-4-en-3-one					
	Crystals	<sup>8</sup> H/ <sup>1</sup> 4C	ML	$^{3}H/^{14}C$	Crystals	³H/¹⁴C	ML	³H/¹⁴ <b>C</b>		
1	1940	1.30	7600	4.90	1720	1.13	1750	1.14		
2	1680	1.14	2580	1.70	1690	1.13	1710	1.13		
3	1700	1.13	1730	1.14						
Calcd <sup>b</sup>					1710					

<sup>&</sup>lt;sup>a</sup> An aliquot of [4-14C]15α-hydroxyprogesterone containing  $1.20 \times 10^5$  dpm was acetylated with [<sup>8</sup>H]acetic anhydride, solution 1, and the product was mixed with 65.0 mg of carrier 15α-acetoxyprogesterone. The mixture was chromatographed on a 7-g alumina column prior to crystallization. <sup>b</sup> A portion of the third crystals (32.5 mg) was reduced with NaBH<sub>4</sub> and oxidized with DDQ, and the product was chromatographed on a 4-g alumina column to yield 21.0 mg of material containing  $3.59 \times 10^4$  dpm of <sup>8</sup>H. The calculated specific activity was determined from these values.

supplied by Dr. P. A. Diassi of the Squibb Institute for Medical Research, who also suggested the method of incubation. The details of this procedure have been described recently (Stern et al., 1968). Following incubation of 0.2 mCi of [4-14C]progesterone (specific activity 37.2 mCi/mmole, New England Nuclear Corp.) with penicillium, a methylene chloride extract was prepared. The extract contained  $4.07 \times 10^8$ dpm and it was chromatographed on paper in system A for 8 hr. The area corresponding in mobility to  $15\alpha$ -hydroxyprogesterone was eluted to yield 2.8 mg of residue containing  $1.05 \times 10^8$  dpm. Chromatography of this residue on paper in system C for 5 hr gave one band of radioactive material with the mobility of  $15\alpha$ -hydroxyprogesterone which on elution afforded a residue containing  $9.00 \times 10^7$  dpm. An aliquot of this residue containing  $1.12 \times 10^5$  dpm was mixed with carrier  $15\alpha$ -hydroxyprogesterone and the mixture was crystallized to constant specific activity as shown in Table II. The determined and calculated specific activities were in good agreement, indicating a high degree of radiochemical purity and the yield of labeled  $15\alpha$ -hydroxyprogesterone was found to be 20%.

Another aliquot of  $[4-14C]15\alpha$ -hydroxyprogesterone containing 1.20 × 105 dpm was acetylated with [3H]acetic anhydride, solution 1. The product was mixed with 65.0 mg of carrier  $15\alpha$ -acetoxyprogesterone, and the mixture was chromatographed on a 7-g alumina column. Elution with benzene-Skellysolve B (4:1) yielded material which was crystallized from acetone-Skellysolve B, ether-Skellysolve B, and acetone, and the specific activities measured are shown in Table III. A portion of the last crystals (32.5 mg) was reduced with 7.5 mg of NaBH4 in 90 ml of methanol. The products were dissolved in 2.5 ml of dioxane and oxidized with 27.5 mg of DDQ. The product obtained was chromatographed on a 4-g alumina column and the material eluted was crystallized twice from acetone-Skellysolve B and the specific activities were determined as shown in Table III. The specific activity of  $[4-14C]15\alpha$ -hydroxyprogesterone was found to be  $3.66 \times 10^8$  dpm/mg calculated from the specific activity of the acetic anhydride and the 3H:14C ratio of the second derivative.

Preparation of Nonlabeled 15 $\alpha$ -Hydroxyprogesterone. When

these studies were started it was necessary to prepare large amounts of  $15\alpha$ -hydroxyprogesterone. This was accomplished by incubating 40 g of progesterone with penicillium, using the conditions described above for the  $15\alpha$  hydroxylation of [4-14C]progesterone, with the exception that the reaction was carried out in a large fermentor (Ayerst Laboratories, Montreal) and a concentration of 400 mg of substrate/l. of incubation medium was employed. The methylene chloride extract of the incubation media weighed 35 g and when crystallized directly from acetone-hexane yielded 11.2 g of crystalline material, mp 208-222°. This material was chromatographed on thin-layer plates (860 plates of 1-mm thickness) using system H. The ultraviolet-positive band corresponding in mobility to  $15\alpha$ -hydroxyprogesterone was eluted from the plates to yield 8.12 g of material which was crystallized twice from acetone to give 6.45 g of coarse needles: mp 230-232°, authentic  $15\alpha$ -hydroxyprogesterone, mp  $231-232^{\circ}$ , mmp 231-232°. The infrared spectrum (KBr) of the product was identical with that of  $15\alpha$ -hydroxyprogesterone.

#### Experimental Section and Results

In all studies to be described the labeled  $15\alpha$ -hydroxyprogesterone was dissolved in 0.5 ml of absolute ethanol and this solution was diluted with 10 ml of isotonic saline prior to intravenous injection. There were insignificant amounts of radioactivity left in the syringe, needle, and vial used in the preparation and administration of the labeled steroids.

Metabolism of  $[7^{-3}H]15\alpha$ -Hydroxyprogesterone by the Normal Male. A total of  $2.3 \times 10^7$  dpm of  $[7^{-3}H]15\alpha$ -hydroxyprogesterone was injected intravenously in a normal 28-year-old male, and on the same day 800 mg of  $15\alpha$ -hydroxyprogesterone, in 80-mg doses contained in gelatin capsules, was ingested over a period of 10 hr. Urine was collected for 6 days and the steroid conjugates in the individual urines were hydrolyzed with Glusulase. A neutral extract was prepared from each day's urine, and these contained a total of 70.5% of the injected dose. Almost all of the radioactive material excreted (99.7%) was present in the urine collected during the first 4 days. As a result, only these extracts were used in the isolation studies.

26 24 20 2 16 DPM/Fraction 12 600 900 100 200 300 500 700 Fraction Number ~ R 10 12

Ethanol in Methylene Chloride

FIGURE 1: Silica gel column chromatography of the neutral extract obtained following Glusulase hydrolysis of the urinary conjugates after the intravenous injection of  $[7^{-3}H]15\alpha$ -hydroxyprogesterone and the oral administration of 800 mg of carrier steroid to a normal male.

The combined neutral extract (1.12 g and 1.62  $\times$  10<sup>7</sup> dpm) was chromatographed on a 150-g silica gel column using increasing concentrations of ethanol in methylene chloride, as previously described (Ruse and Solomon, 1966). The effluent from the column was collected in 10-ml fractions at the rate of 30-40 ml/hr and seven major pools of radioactive material were eluted as shown in Figure 1. Owing to the small weight of material present in pools A, E, and G after further chromatography, the metabolites eluted in these fractions could not be identified. The residue from pool B (80 mg and  $3.94 \times 10^6$ dpm) was chromatographed on a 100-g Celite column using system D and two major pools of radioactive material, BI and BII, were eluted in the 4th and 5-7 hold-back volumes, respectively. BI (14.1 mg and  $4.28 \times 10^5$  dpm) was chromatographed on paper in system A for 12 hr and then on a 1-g silica gel column from which 3.3 mg of material containing  $2.99 \times 10^5$  dpm was eluted with 3% ethanol in methylene chloride. Crystallization of this material from acetonemethanol gave 0.5 mg of coarse needles. The infrared spectrum (KBr) of the crystals showed a strong hydroxyl band at 3450 cm<sup>-1</sup> and a band at 1695 cm<sup>-1</sup> indicating the possible presence of a 20-ketone. Because of the small amount isolated the identity of this unknown could not be established.

Residue BII weighed 53.6 mg and contained  $3.93 \times 10^8$  dpm. Chromatography of a small aliquot of BII on paper in system A for 10 hr indicated that it contained a single band of radioactive material identical in mobility to  $15\alpha$ -hydroxy-progesterone. Therefore the remainder of BII (50.9 mg) was chromatographed on a 5-g alumina column and 43.3 mg of crystalline material containing  $3.62 \times 10^6$  dpm was eluted with 1% ethanol in benzene. Crystallization of this material from acetone gave 27.4 mg of coarse needles: mp 230–232°, specific activity  $1.08 \times 10^5$  dpm/mg. The infrared spectrum (KBr) of the crystals was identical with that of authentic  $15\alpha$ -hydroxyprogesterone.

The material present in pool C (Figure 1, 81 mg and  $4.52 \times 10^6$  dpm) was chromatographed on a 100-g Celite column using system E and a single peak containing 56 mg of material and  $4.48 \times 10^6$  dpm was eluted in the 4th hold-back volume. This material was further purified by chromatography on paper in system A for 24 hr, and on a 6-g silica gel column

from which 51.4 mg of an oily residue containing 4.45  $\times$ 10<sup>6</sup> dpm was eluted with 4% ethanol in methylene chloride. It was crystallized with great difficulty from acetone-ether-Skellysolve B to yield 14.0 mg of fine plates: mp 175-177°, specific activity 1.42 × 10<sup>5</sup> dpm/mg. An aliquot (8.3 mg) was acetylated with [14C]acetic anhydride and the product was chromatographed on alumina and 10.2 mg of oily residue was eluted from the column with benzene-Skellysolve B (3:2). Crystallization of the acetate from acetone-Skellysolve B afforded 6.5 mg of coarse plates, mp 142-144°. The specific activities of the crystals and the mother liquors were both  $4.08 \times 10^4$  dpm of  ${}^{14}$ C/mg ( ${}^{3}$ H/ ${}^{14}$ C = 1.98). These results indicated that the material in pool C had two acylable hydroxyl groups. Its infrared spectrum taken in KBr (Figure 2a) showed a hydroxyl band at 3350 cm<sup>-1</sup>, a carbonyl band at 1705 cm<sup>-1</sup> corresponding to a 20-ketone, and a band at 1002 cm<sup>-1</sup> which is compatible with the  $3\alpha$ - $5\alpha$  stereochemistry in the A ring (Cole et al., 1952). The spectrum (CS<sub>2</sub>) of the acetate indicated the absence of a hydroxyl band, the presence of a 20-ketone at 1710, and a band at 1018 cm<sup>-1</sup> indicative of  $3\alpha$ -acetoxy with an axial conformation (Jones and Herling, 1956). The mass spectrum gave a mass number of 334, which was consistent with the structural formula  $C_{21}H_{34}O_{3}$ . Table IV shows the results obtained from nuclear magnetic resonance analysis of the alcohol and its diacetate.

The chemical shifts of the angular methyls (C-18 and C-19) indicate that the compound is of the  $5\alpha$  rather than the  $5\beta$ series of steroids, and agree well with values calculated from Zürcher's Tables (Zürcher, 1963) for  $3\alpha$ ,  $15\alpha$ -dihydroxy- $5\alpha$ pregnan-20-one and its diacetate (Table IV). These same methyl signals differed sufficiently from those calculated for the corresponding  $3\alpha,15\beta$ - and  $3\beta,15\beta$ -diols and their diacetates which confirmed that the substituent at 15 remained in the  $\alpha$  configuration. However, the differences between a  $3\alpha,15\alpha$  and  $3\beta,15\alpha$  substituent were not so great as to be conclusive, although agreement with the values calculated for the  $3\alpha$  isomer was somewhat better than for the  $3\beta$ . Fortunately this difference is easily distinguished by the spin-spin splitting patterns of the proton at the C-3 position. In the  $5\alpha$  system, a substituent in the  $3\alpha$  position is accompanied by an equatorial proton, while a  $3\beta$  substituent has

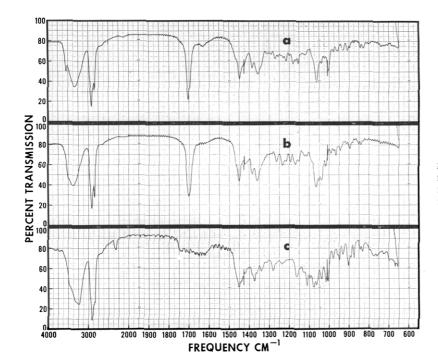


FIGURE 2: Infrared spectra (KBr): (a) metabolite isolated from pool C, (b) metabolite isolated from pool D, and (c) metabolite isolated from pool F.

an axial proton at C-3. In the absence of further substitution at positions 2 and 4, an axial proton at C-3 has two axial-axial coupling interactions (large) and two axial equatorial couplings (small to medium). A similarly placed equatorial proton has two equatorial-axial and two equatorial-equatorial couplings (all small to medium). Thus the axial proton may be distinguished by its broader coupling pattern.

The nuclear magnetic resonance spectrum of C had two proton signals (C-3-H and C-15-H) which were superimposed in the region 3.9–4.2 ppm in the diol and 4.8–5.1 ppm in the diacetate. The appearance was one of a sharper signal atop a much broader one. Identification of the sharper signal as that of C-3-H and thus assignment of this proton as equatorial was accomplished by comparing the low-field region of the nmr of compound C with that of D (vide infra) in which two very broad signals occurred. (The fact that D belongs to the  $5\beta$  series will have little or no effect upon the splitting pattern of its  $15\alpha$ -H.) From the nuclear magnetic resonance results combined with the information obtained from the infrared spectra, the number of acylable hydroxyl groups and from the mass spectrum, the structure  $3\alpha$ ,  $15\alpha$ -dihydroxy- $5\alpha$ -pregnan-20-one can be assigned to the metabolite isolated from pool C.

The residue from pool D (Figure 1) weighed 117 mg and contained  $3.94 \times 10^6$  dpm. Chromatography of this residue on a 100-g Celite column in system E yielded 60.3 mg of material containing  $3.62 \times 10^6$  dpm which was eluted in the 5th and 6th hold-back volume. This material was further purified by chromatography on paper in system A for 24 hr and on a 6-g silica gel column from which 43.0 mg of crystalline residue containing  $3.33 \times 10^6$  dpm was eluted with 4% ethanol in methylene chloride. Crystallization of this material from acetone–methanol–Skellysolve B afforded 23.0 mg of coarse plates: mp 208–210°, specific activity  $7.90 \times 10^4$  dpm/mg. An aliquot (9.3 mg) of the crystals was acetylated with [14C]acetic anhydride and the product was chromatographed on an alumina column and elution with benzene–Skellysolve

B (3:2) yielded 12 mg of colorless oil. Crystallization of this material from acetone-Skellysolve B afforded 10.9 mg of fine needles, mp 164-166.5°. The specific activities of the crystals and the mother liquors were both  $3.89 \times 10^4$  dpm of  ${}^{14}\text{C/mg}$  ( ${}^{3}\text{H}/{}^{14}\text{C} = 1.65$ ). These results indicated the presence of two acylable hydroxyl groups in the unknown in pool D. The infrared spectrum of this material (Figure 2b) showed the presence of a hydroxyl band at 3375, a carbonyl band at 1705 corresponding to a 20-ketone, and a band at 1038 cm<sup>-1</sup> corresponding to an equatorial  $3\alpha$ -hydroxyl (Cole et al., 1952). The infrared spectrum (CS2) of the acetate had no hydroxyl band but had a 1710-cm<sup>-1</sup> band, indicating the presence of a 20-ketone and a band at 1029 cm<sup>-1</sup> corresponding to an equatorial  $3\alpha$ -acetoxy group (Jones and Herling, 1956). The mass spectrum indicated that the unknown in pool D had a molecular weight of 334, which was consistent with the structural formula C21H34O3.

Table IV shows the results obtained from nuclear magnetic resonance analysis of unknown D and its diacetate. The chemical shifts of the angular methyls identify this compound as being in the  $5\beta$  series, and agree well with values calculated from Zürcher's tables (Zürcher, 1963) for  $3\alpha$ ,  $15\alpha$ -dihydroxy- $5\beta$ -pregnan-20-one and its diacetate. The calculated values for the  $3\alpha,15\beta$ - and  $3\alpha,15\beta$ -diols and their diacetates were sufficiently different as to indicate that no change had occurred at the 15 position. The calculated values did not distinguish unambiguously between  $3\alpha$  and  $3\beta$ . However, two very broad multiplets at approximately 3.63 and 4.02 ppm in the diol which shifted together into the region 4.5 to 5.0 ppm in the diacetate identified the C-3 proton as axial (very broad coupling pattern described above). The nuclear magnetic reasonance data therefore indicated that the unknown D was  $3\alpha$ ,  $15\alpha$ -dihydroxy- $5\beta$ -pregnan-20-one. This conclusion was reinforced by the data obtained from the infrared spectra, the number of acylable hydroxyl groups, and from the mass spectrum. Confirmatory evidence for the assignment of this

TABLE IV: Nuclear Magnetic Resonance Data for Metabolites in Pools C, D, and F.a

Metabolite	C-19 CH <sub>3</sub>	C-21 C-18 CH <sub>3</sub> CH <sub>3</sub>		CH₃COO	C-3 H	Confirmation of C-3 H	
С	0.79	0.63	2.10		3.9-4.2	Equatorial	
C-diacetate	0.79	0.68	2.10	1.98, 2.03	4.8-5.1	Equatorial	
D	0.93	0.64	2.10		3.6-4.1	Axial	
D-diacetate	0.94	0.67	2.10	1.98, 2.02	4.5-4.9	Axial	
F-triacetate	0.80	0.72	1.216	1.98, 1.99, 2.04	4.95		
3α,15α-Dihydroxy-5α- pregnan-20-one	0.792°	0.650°					
$3\alpha$ , $15\alpha$ -Diacetoxy- $5\alpha$ -pregnan-20-one	0.809°	0.693°					
$3\alpha$ , $15\alpha$ -Dihydroxy- $5\beta$ -pregnan-20-one	0.933	0 . 650°					
$3\alpha$ , $15\alpha$ -Diacetoxy- $5\beta$ -pregnan-20-one	0.942°	0.684					
$5\alpha$ -Pregnane $3\alpha$ , $15\alpha$ , $20\alpha$ - triacetate	0.809°	0.743°					
$5\alpha$ -Pregnane $3\beta$ , $15\alpha$ , $20\alpha$ - triacetate	0.834	0.734					
$5\alpha$ -Pregnane $3\alpha$ , $15\alpha$ , $20\beta$ - triacetate	0.809°	0.703°					
$5\alpha$ -Pregnane $3\beta$ , $15\alpha$ , $20\beta$ - triacetate	0.834	0.674°					

<sup>&</sup>lt;sup>a</sup> The spectra were measured with a Varian HR-100 spectrometer. Peak positions are expressed in parts per million (ppm) relative to a tetramethylsilane internal standard. The audio-side-band method of calibration was used. <sup>b</sup> Doublet, J = 6 Hz. <sup>c</sup> These values were calculated for the compounds designated using Zürcher's tables (Zürcher, 1963).

structure was derived from the specific stereochemical reduction of  $15\alpha$ -hydroxyprogesterone with *Clostridium paraputrificum* (Schubert *et al.*, 1965) through the courtesy of Dr. Josef Schlegel, Deutsche Akademie der Wissenschaften zu Berlin, Jena, Germany. The infrared spectrum (KBr) of the reduced product sent by Dr. Schlegel was identical with that of  $3\alpha$ ,  $15\alpha$ -dihydroxy- $5\beta$ -pregnan-20-one except for a shoulder at 1670 cm<sup>-1</sup> indicating the presence of a small amount of unreacted substrate in the sample.

The residue obtained from pool F (29.0 mg and 6.60  $\times$ 105 dpm) was purified by chromatography on paper in system A for 4 days and then on a 2-g silica gel column from which 9.0 mg of oily material containing 6.15  $\times$  10 $^{5}$  dpm was eluted with 10% ethanol in methylene chloride. Crystallization of this material from acetone-Skellysolve B yielded 1.6 mg of small plates: mp 211-214°, specific activity 1.96  $\times$ 10<sup>5</sup> dpm/mg. An aliquot (0.95 mg) was acetylated with [14C]acetic anhydride and the product was chromatographed on alumina. Elution from the column with benzene-Skellysolve B (1:1) yielded 1.2 mg of colorless oil which had a specific activity of  $5.52 \times 10^4$  dpm of  $^{14}$ C/mg ( $^{3}$ H/ $^{14}$ C = 3.1). These results indicated the presence of three acylable hydroxyl groups in the unknown F. The infrared spectrum of F (Figure 2c) showed a hydroxyl band at 3250 cm<sup>-1</sup> and no carbonyl bands indicating that both the  $\Delta^4$ -3-keto and the 20-keto groups of  $15\alpha$ -hydroxyprogesterone had been reduced during the course of metabolism. The infrared spectrum (CS<sub>2</sub>) of the triacetate of F had no hydroxyl band, and a broad acetate band was observed at  $1240 \text{ cm}^{-1}$ . The mass spectrum indicated that the unknown in pool F had a molecular weight of 336, which was consistent with the structural formula  $C_{21}H_{36}O_3$ .

Table IV shows the results obtained from nuclear magnetic resonance analysis of the triacetate of metabolite F. Instead of a methyl ketone at C-21 (2.12 ppm) the spectrum showed a doublet at 1.20 ppm (J = 7 Hz) due to a secondary methyl group formed by reduction of the C-20 ketone to an alcohol. The positions of the angular methyls corresponded to those of a  $5\alpha$ -steroid, but were similar enough to the calculated values for both  $5\alpha$ -pregnane- $3\alpha$ ,  $15\alpha$ ,  $20\alpha$ -triol and the  $20\beta$ triol triacetates that it was not possible to determine whether the reduction of the keto group gave the  $20\alpha$ - or  $20\beta$ -alcohol. The observed values were in somewhat better agreement for the  $3\alpha$ - rather than  $3\beta$ -acetate, though this difference was also small. In this case the signals of the three protons adjacent to the acetate groups (C-3, C-15, and C-20) overlapped badly and it was not possible to identify the splitting pattern of the proton at C-3.

Following reduction of an aliquot of the diacetate of metabolite C,  $3\alpha,15\alpha$ -diacetoxy- $5\alpha$ -pregnan-20-one, with NaBH<sub>4</sub> and acetylation of the product, a triacetate was obtained whose infrared spectrum (CS<sub>2</sub>) was identical with that of F-triacetate. Since it is known that NaBH<sub>4</sub> reduction

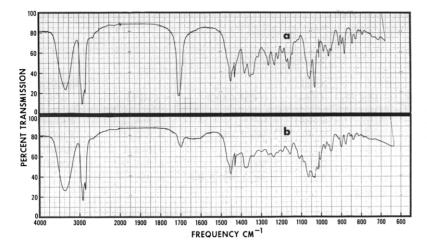


FIGURE 3: Infrared spectra (KBr): (a) metabolite isolated from pool II and (b) metabolite isolated from pool VI.

of a C-20 ketone favors the formation of the  $20\beta$ -alcohol (Norymberski and Woods, 1955) the unknown F most probably was  $5\alpha$ -pregnane- $3\alpha$ ,  $15\alpha$ ,  $20\beta$ -triol.

To confirm this it was necessary to study the possible effect of the 15α-hydroxyl group on the NaBH<sub>4</sub> reduction of 20ketosteroids. Fieser and Fieser (1948) summarized the molecular rotation, MD, data for 19 pairs of C-20 epimers in the pregnane series and pointed out that the shift in Mp on acetylation is positive for  $20\beta$ -ols and negative for  $20\alpha$ -ols. An aliquot (198 mg) of  $15\alpha$ -acetoxyprogesterone was reduced with NaBH<sub>4</sub>, the product was oxidized with DDQ, and the material thus obtained was chromatographed on an alumina column. Elution with 0.5% ethanol in benzene yielded a product which was crystallized from acetone-Skellysolve B to give 115 mg of crystals. An aliquot of this material (50 mg) was acetylated and the product was crystallized from methanol-ether. Aliquots of the 20-alcohol and its acetate (30 and 31 mg, respectively) were used for optical rotation studies. The optical rotation of  $20\alpha$ -dihydroprogesterone,  $20\beta$ dihydroprogesterone, and their respective acetates were also measured. Table V shows the molecular rotation difference,

TABLE V: Optical Rotation Data.

Steroid	Molecular Rotation (MD) <sup>a</sup>	$\Delta M$ D
20α-Dihydroprogesterone	-306	
$20\alpha$ -Dihydroprogesterone acetate	-312	-6
20β-Dihydroprogesterone	-363	
20β-Dihydroprogesterone acetate	-151	+212
Product obtained from the NaBH <sub>4</sub>	-242	·
reduction followed by DDQ oxidatio	n	
of $15\alpha$ -acetoxyprogesterone		
$15\alpha$ , $20\beta$ -Diacetoxyprogesterone <sup>b</sup>	-63	+179

 $<sup>^</sup>a$  Optical rotations were obtained with a Rudolph polarimeter using 1% chloroform solutions.  $^b$  The diacetate was obtained by reduction of 15 $\alpha$ -acetoxyprogesterone with NaBH<sub>4</sub>, oxidation of this product with DDQ reagent, and subsequent acetylation.

 $\Delta$ MD, obtained after acetylation of  $20\alpha$ -dihydroprogesterone and  $20\beta$ -dihydroprogesterone. The same correlation reported by Fieser and Fieser (1948) was observed here in that the  $\Delta MD$  of the  $20\alpha$  epimer was slightly negative (-6°) while the  $\Delta MD$  of the 20 $\beta$  epimer was positive (+212°). As shown in Table V the  $\Delta MD$  due to the introduction of a 20-acetoxy in  $15\alpha$ -acetoxy-20-hydroxypregn-4-en-3-one was +179. These results provide confirmatory evidence that NaBH4 reduction of 20-ketosteroids, which have a  $15\alpha$ -hydroxy, results in the formation of 20β-alcohols. Further confirmation was obtained by the reduction of  $15\alpha$ -acetoxyprogesterone with 20β-hydroxysteroid dehydrogenase as described in the Methods section. The infrared spectra (CS<sub>2</sub>) of the product obtained after reduction of  $15\alpha$ -acetoxyprogesterone with  $20\beta$ -hydroxysteroid dehydrogenase in the presence of  $\beta$ -DPNH was identical with the spectrum of the compound obtained after NaBH4 reduction followed by DDQ oxidation. Since reduction of  $3\alpha$ ,  $15\alpha$ -diacetoxy- $5\alpha$ -pregnan-20-one with NaBH<sub>4</sub> and acetylation of the product yielded material having an infrared spectrum identical with the F-triacetate, the structure  $5\alpha$ -pregnane- $3\alpha$ ,  $15\alpha$ ,  $20\beta$ -triol could be assigned to unknown F.

In order to secure large amounts of nonlabeled metabolites of  $15\alpha$ -hydroxyprogesterone for use as carriers in the investigations to be described the following study was performed. Each of four normal male volunteers was given 800 mg of  $15\alpha$ hydroxyprogesterone by mouth, in 80-mg doses contained in gelatin capsules, over a period of 10 hr. A total of 3.25 g of  $15\alpha$ -hydroxyprogesterone was administered to the four subjects. The urine collected from each subject for 4 days was pooled and the urinary conjugates were hydrolyzed with Glusulase. The neutral extract obtained after hydrolysis (3.5 g) was chromatographed on a 250-g silica gel column using increasing concentrations of ethanol in methylene chloride and the effluent from the column was collected in 20-ml fractions at a rate of 50-60 ml/hr. Since no radioactive steroid was administered in this study, the fractions obtained from the column were pooled on the basis of visual examination of the tubes after evaporation of the solvent. Six pools containing crystalline material were obtained as shown in Table VI.

From pools I, III, and IV, crystalline  $15\alpha$ -hydroxyprogesterone (190 mg),  $3\alpha$ , $15\alpha$ -dihydroxy- $5\alpha$ -pregnan-20-one (110 mg), and  $3\alpha$ , $15\alpha$ -dihydroxy- $5\beta$ -pregnan-20-one (320 mg), re-

TABLE VI: Combined Fractions from the Initial Silica Gel Column of the Neutral Extract of Urine Following the Ingestion of  $15\alpha$ -Hydroxyprogesterone.

		% Ethanol in Methylene			
Pool	Fraction	Chloride	Weight (mg)	Metabolite Isolated	
I	174-230	3	280	15α-Hydroxyprogesterone	
II	281-320	3	110	$3\beta$ , $15\alpha$ -Dihydroxy- $5\beta$ -pregnan-20-one	
III	321-440	4	230	$3\alpha$ , $15\alpha$ -Dihydroxy- $5\alpha$ -pregnan-20-one	
IV	441-530	4–6	460	$3\alpha,15\alpha$ -Dihydroxy- $5\beta$ -pregnan-20-one	
V	786-820	10	42	$5\alpha$ -Pregnane- $3\alpha$ , $15\alpha$ , $20\beta$ -triol	
VI	881-995	12	110	$5\beta$ -Pregnane- $3\alpha$ , $15\alpha$ , $20\beta$ -triol	

TABLE VII: Nuclear Magnetic Resonance Data for Urinary Metabolites II and VI<sup>a</sup> Following Ingestion of  $15\alpha$ -Hydroxyprogesterone by Normal Males.

Metabolite	C-19 CH <sub>3</sub>	C-18 CH₃	C-21 C CH <sub>3</sub>	CH₃COO	С-3 Н	Confirmation of C-3 H
II	0.99	0.64	2.12		3.9-4.2	Equatorial
II-diacetate	0.98	0.68	2.12	2.00, 2.05	4.8-5.2	Equatorial
VI-acetate	0.93	0.71	1.216	1.99, 2.00, 2.02	4.7-5.0	•
$3\beta$ , $15\alpha$ -Dihydroxy- $5\beta$ -pregnan-20-one	0.975°	0.650°		, ,		
$3\beta$ , $15\alpha$ -Diacetoxy- $5\beta$ -pregnan-20-one	0.975°	0.684°				
$5\beta$ -Pregnane $3\alpha$ , $15\beta$ - $20\alpha$ -triacetate	0.942°	0. <b>734</b> °				
$5\beta$ -Pregnane $3\alpha$ , $15\alpha$ , $20\beta$ -triacetate	0.942°	0. <b>684</b> °				

<sup>&</sup>lt;sup>a</sup> The spectra were measured with a Varian HR-100 spectrometer and were displayed on precalibrated charts. Peak positions are expressed in parts per million (ppm) relative to a tetramethylsilane internal standard. <sup>b</sup> Doublet, J = 6 Hz. <sup>c</sup> These values were calculated for the compounds designated using Zürcher's tables (Zürcher, 1963).

spectively, were isolated and identified by the methods previously described. From pool II a residue was obtained (110 mg) which was chromatographed on 7 thin-layer chromatography plates using system H and a product migrating as a single band giving a positive phosphomolybdic acid reaction was obtained at an  $R_F$  of 0.45 (II). The material eluted weighed 95 mg and it was then chromatographed on a 10-g alumina column. Elution with 1.5% ethanol in benzene yielded 83 mg of crystalline residue which on crystallization from acetone-Skellysolve B afforded 45 mg of coarse plates, mp 192-195°. An aliquot (2.0 mg) of II was acetylated with [14C]acetic anhydride and the product was chromatographed on an alumina column from which 2.1 mg of colorless oil was eluted with benzene-Skellysolve B (3:2). Crystallization of this material from acetone-Skellysolve B afforded 1.6 mg of fine needles, mp 152-155°. The specific activities of the crystals and the mother liquors were both  $3.80 \times 10^4 \, \mathrm{dpm}/$ mg. From this specific activity it was possible to calculate that the unknown II had two acylable hydroxyl groups. The

infrared spectrum (KBr) of the unknown II (Figure 3a) showed a hydroxyl band at 3350 cm<sup>-1</sup>, a carbonyl band at 1710 cm<sup>-1</sup> indicating the presence of a 20-ketone, and a band at 1032 cm<sup>-1</sup> corresponding to an axial  $3\beta$ -hydroxyl (Cole et al., 1952). The infrared spectrum of the acetate (CS2) indicated the absence of a hydroxyl band, a band at 1710 cm-1 corresponding to a 20-ketone, and a band at 1022 cm<sup>-1</sup> compatible with the presence of an axial  $3\beta$ -acetoxy (Jones and Herling, 1956). The mass spectrum indicated that 11 had a molecular weight of 334, which was consistent with the structural formula C21H34O3. Table VII shows the results obtained from the nuclear magnetic resonance analysis of unknown II and its diacetate. The chemical shifts of the angular methyls (C-18, C-19) of the diol II and its diacetate corresponded to those of the  $5\beta$ -steroid series. The presence of a three proton singlet at 2.12 ppm indicated that the 17acyl group was still intact. The protons at C-3 and C-15 overlapped badly in the diol, but separated enough to be discernible in the diacetate. The signal of the 3 proton

FIGURE 4: Structural formulas of the urinary metabolites of  $15\alpha$ -hydroxyprogesterone.

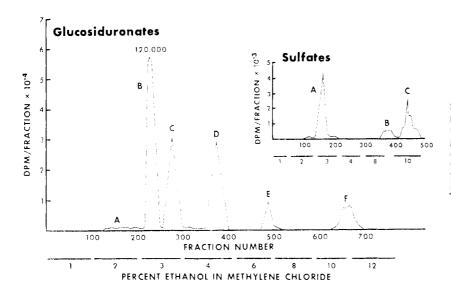


FIGURE 5: Silica gel column chromatography of the neutral extracts obtained following solvolysis (sulfates) and  $\beta$ -glucuronidase hydrolysis (glucosiduronates) of the urinary conjugates after the intravenous injection of [4-14C]15 $\alpha$ -hydroxyprogesterone into a normal subject in the 38th week of pregnancy.

appeared at 5.09 ppm, and at half-height had a width of 9 to 10 Hz while that of the 15 proton appeared at 4.89 ppm and was widest at 20 Hz, thus only partly overlapping the 3-proton signal. This permitted assignment of C-3-H as equatorial by comparison with the spectrum of D-diacetate (previous study) in which the only difference is at C-3. These results permit the assignment of the structure  $3\beta$ ,15 $\alpha$ -dihydroxy-5 $\beta$ -pregnan-20-one to the metabolite isolated from pool II.

From pool VI a residue was obtained which weighed 110 mg. It was purified by chromatography on eight thin-layer chromatography plates using system J, and then on a 5-g silica gel column to yield 18.5 mg of oily material (VI) which could not be crystallized. Acetylation of an aliquot (12.0 mg) of VI and chromatography on an alumina column yielded 8.0 mg of a crystalline residue. Crystallization of this residue from acetone–Skellysolve B afforded 4.5 mg of small plates, mp 206–208°. The infrared spectrum of VI (Figure 3b) showed a hydroxyl band at 3350 cm<sup>-1</sup> and no carbonyl bands indicating that both the  $\Delta^4$ -3-keto and the 20-keto groups of the administered  $15\alpha$ -hydroxyprogesterone had been reduced. The infrared spectrum (CS<sub>2</sub>) of the acetate of

VI had no hydroxyl band, and a strong acetate band was observed at 1240 cm $^{-1}$ . The mass spectrum of VI indicated that it had a molecular weight of 336, which was consistent with the structure  $C_{21}H_{36}O_3$ .

The results obtained from nuclear magnetic resonance analysis of VI-acetate are shown in Table VII. The positions of the angular methyl groups placed this compound in the  $5\beta$  series along with II and its diacetate. The presence of a doublet at 1.21 ppm (J = 6 Hz) and an additional proton in the low-field region of the spectrum (along with signals of H-3 and H-15) indicated that the carbonyl had been reduced to an alcohol and that the three methyl signals in the vicinity of 2 ppm were all acetate methyls. The calculated values of the angular methyls for both the  $20\alpha$ - and  $20\beta$ -acetates were quite close to the observed ones, thus it was not possible to distinguish between these two structures by nuclear magnetic resonance. Similarly, the differences calculated for the  $3\alpha$ and  $3\beta$ -acetates were too small to differentiate between the two structures on that basis alone. With the three overlapping proton signals in the same region, it is difficult to ascertain whether the C-3-H is axial or equatorial, but the data appear more consistent with its being axial and with an equatorial

TABLE VIII: Purification of Metabolites Isolated from Urine Following the Intravenous Injection of  $[4-14C]15\alpha$ -Hydroxyprogesterone in a Subject in the 38th Week of Pregnancy.

Radioactive Material Eluted from Silica Gel Column	dpm of ¹4C Eluted	Wt (mg)	Procedure Used for Proof of Radiochemical Sequence of Chromatography Purity Metabolite Identifi					
	- · · · · · · · · · · · · · · · · · · ·		Sulfates					
Α	$8.70 \times 10^{4}$	35.6	$Pc - A^a \rightarrow$					
			$Pc - G \rightarrow$		$3\beta$ , $15\alpha$ -Dihydroxy- $5\beta$ -			
			Sg <sup>b</sup> → acetylation <sup>c</sup>	Rid. d	pregnan-20-one			
			Glucosiduronates		-			
В	$1.36  imes 10^6$	30.0	$Pc - A \rightarrow PC - B \rightarrow$		$15\alpha$ -Hydroxyprogesterone			
			Sg → acetylation	Rid.				
C	$3.07 \times 10^{5}$	60.0	$Pc - A \rightarrow Pc - G \rightarrow$		$3\alpha$ , $15\alpha$ -Dihydroxy- $5\alpha$ -			
			$Sg \rightarrow acetylation$	Rid.	pregnan-20-one			
D	$6.81  imes 10^{5}$	64.0	$Pc - A \rightarrow Pc - G \rightarrow$		$3\alpha$ , $15\beta$ -Dihydroxy- $5\beta$ -			
			Sg → acetylation	Rid.	pregnan-20-one			
F	$1.55 \times 10^{5}$	21.0	Pc - A					
FI۰	$4.20 \times 10^{4}$	4.5			$5\alpha$ -Pregnane- $3\alpha$ , $15\alpha$ , $20\beta$ -			
FII	$1.05 \times 10^{5}$	5.2	$Pc - G \rightarrow Sg \rightarrow acetylation$	Rid.	triol			

 $^a$  Pc - A, paper chromatography using system A.  $^b$  Sg, silica gel chromatography.  $^c$  Acetylation with [ $^3$ H]acetic anhydride solution 1.  $^d$  Rid., reverse isotope dilution.  $^e$  The material in this fraction was not identified.

acetate. However, when an aliquot of  $3\alpha,15\alpha$ -diacetoxy- $5\beta$ -pregnan-20-one was reduced with NaBH<sub>4</sub> and the product acetylated, the resulting triacetate had an infrared spectrum almost identical with that of VI-triacetate. The only difference between the two spectra was an extra band at  $1145 \text{ cm}^{-1}$  in the spectrum of VI-triacetate, which might be due to the presence of impurities in this compound since only small amounts of VI-triacetate were available and the material could only be crystallized once. Because it was established that NaBH<sub>4</sub> reduction of  $15\alpha$ -hydroxylated 20 ketosteroids resulted in the formation of  $20\beta$ -alcohols, metabolite VI was assigned the structure of  $5\beta$ -pregnane- $3\alpha,15\alpha,20\beta$ -triol.

In the two studies described above, it was possible to isolate and identify the following urinary metabolites:  $15\alpha$ -hydroxyprogesterone,  $3\alpha$ , $15\alpha$ -dihydroxy- $5\alpha$ -pregnan-20-one,  $5\alpha$ -pregnane- $3\alpha$ , $15\alpha$ -dihydroxy- $5\beta$ -pregnan-20-one,  $5\alpha$ -pregnane- $3\alpha$ , $15\alpha$ , $20\beta$ -triol,  $3\beta$ , $15\alpha$ -dihydroxy- $5\beta$ -pregnan-20-one, and  $5\beta$ -pregnane- $3\alpha$ , $15\alpha$ , $20\beta$ -triol. The structure of these metabolites are shown in Figure 4.

Metabolism of  $[4^{-14}C]15\alpha$ -Hydroxyprogesterone by the Pregnant Female. A total of 13.4  $\mu$ g of  $[4^{-14}C]15\alpha$ -hydroxyprogesterone containing  $4.90 \times 10^6$  dpm (specific activity  $3.66 \times 10^5$  dpm/ $\mu$ g) was injected intravenously in a normal volunteer in the 38th weeks of pregnancy. Urine was collected for 6 days and the conjugates present in each day's urine were first hydrolyzed by sovolysis and then with  $\beta$ -glucuronidase. It was found that 6.8% of the injected radioactivity was excreted in the sulfate fraction and 66.1% in the glucosiduronate fraction of the urine. As was observed in the first study, practically all of this radioactivity was found in the first 4 day's urine and the extracts from this period were used for the isolation studies.

The sulfate fraction (450 mg and  $3.36 \times 10^5$  dpm) was

chromatographed on a 50-g silica gel column and three major pools of radioactivity were obtained as shown in Figure 5. Metabolites eluted in pools B and C from this column have not as yet been identified. The glucosiduronate fraction (978 mg and  $3.24 \times 10^6$  dpm) was chromatographed on a 100-g silica gel column and the radioactive metabolites eluted are shown in Figure 5. Metabolites eluted in pools A and E from this column have not as yet been identified. The procedures used for the purification and identification of the metabolites isolated from the sulfate and glucosiduronate fractions of the urine are outlined in Table VIII. Each pool obtained from the initial silica gel column was purified by chromatography on paper in two systems and the material eluted from the second paper chromatogram was further purified by chromatography on a silica gel column, and then the metabolite was acetylated with [3H]acetic anhydride, solution 1. Following acetylation, the appropriate carrier acetate, obtained in the previous study, was added and the mixture was crystallized to constant specific activity and constant <sup>3</sup>H/<sup>14</sup>C ratio. In all cases except one (metabolite FII, glucosiduronate fraction) a suitable derivative was formed and it was crystallized to constant specific activity and 3H/14C ratio.

A derivative of  $15\alpha$ -acetoxyprogesterone,  $15\alpha$ -acetoxy- $20\beta$ -hydroxy-pregn-4-en-3-one, was prepared as previously described, and derivatives of  $3\beta$ ,  $15\alpha$ -diacetoxy- $5\beta$ -pregnan-20-one,  $3\alpha$ ,  $15\alpha$ -diacetoxy- $5\beta$ -pregnan-20-one were prepared by reduction with NaBH<sub>4</sub>. A second derivative of  $5\alpha$ -pregnane- $3\alpha$ ,  $15\alpha$ -dihydroxy- $5\beta$ -pregnan-20-one was isolated from pool A of the sulfate fraction, and  $15\alpha$ -hydroxyprogesterone,  $3\alpha$ ,  $15\alpha$ -dihydroxy- $5\alpha$ -pregnan-20-one,  $3\alpha$ ,  $15\alpha$ -dihydroxy- $5\alpha$ -pregnan-20-one,  $3\alpha$ ,  $15\alpha$ -dihydroxy- $5\beta$ -pregnan-20-one,

TABLE IX: Specific Activity and Percentage Recovery of Urinary Metabolites from Injected [4-14C]15 $\alpha$ -Hydroxyprogesterone in a Pregnant and a Nonpregnant Subject.<sup>a</sup>

		Pregnant Subj	ect	N	Nonpregnant Subject			
	Specific Activity $^b$ (dpm/ $\mu$ g)		Minimal % of Injected Radio- activity in Urinary	Specific Activity <sup>a</sup> (dpm/µg)		Minimal % of Injected Radio- activity in Urinary		
Metabolite Isolated	Se	G/	Product <sup>o</sup>	S	G	Product		
15α-Hydroxyprogesterone	_h	$1.56 \times 10^{4}$	18.7	_	$3.31 \times 10^{5}$	5.8		
$3\alpha$ , $15\alpha$ -Dihydroxy- $5\alpha$ -pregnan-20-one	<b></b>	$7.90 \times 10^{3}$	2.9	-	$3.45 \times 10^{5}$	7.1		
$3\alpha,15\alpha$ -Dihydroxy- $5\beta$ -pregnan-20-one	-	$9.70 \times 10^{3}$	7.5	-	$3.52 \times 10^5$	7.4		
$3\beta$ , $15\alpha$ -Dihydroxy- $5\beta$ -pregnan-20-one	$8.66 \times 10^{8}$		1.0	$3.06 \times 10^{5}$	-	0.4		
$5\alpha$ -Pregnane- $3\alpha$ , $15\alpha$ , $20\beta$ -triol	-	$8.68 \times 10^{3}$	1.8		$3.35 \times 10^{5}$	2.3		

<sup>&</sup>lt;sup>a</sup> The specific activity of the injected [4-14C]15α-hydroxyprogesterone was  $3.66 \times 10^5$  dpm/µg. <sup>b</sup> These specific activities were calculated from the specific activity of [<sup>3</sup>H]acetic anhydride, solution 1, and the <sup>3</sup>H/<sup>14</sup>C ratios obtained following crystallization of the final derivative. The specific activities and <sup>3</sup>H/<sup>14</sup>C ratios agreed within at least 4%. <sup>c</sup> These values were computed from the final specific activity of the crystals and the weight of carrier added. <sup>d</sup> These specific activities were calculated from the specific activity of [<sup>3</sup>H]acetic anhydride, solution 2, and the <sup>3</sup>H/<sup>14</sup>C ratios obtained following crystallization of the final derivative. The agreement of specific activities and <sup>3</sup>H/<sup>14</sup>C ratios was within 7% for 15α-hydroxyprogesterone and 4% or better for the other metabolites. <sup>e</sup> S, sulfate. <sup>f</sup> G, glucosiduronate. <sup>h</sup> The dashes indicate the metabolite was not isolated.

and  $5\alpha$ -pregnane- $3\alpha$ ,  $15\alpha$ ,  $20\beta$ -triol were isolated from pools B, C, D, and F, respectively, of the glucosiduronate fraction. The specific activities of the isolated metabolites are shown in Table IX. These specific activities were calculated using the specific activity of the [ $^3$ H]acetic anhydride and the final  $^3$ H/ $^1$ C ratios obtained following crystallization of the derivative.

Metabolism of [4-14C]15α-Hydroxyprogesterone by the Nonpregnant Female. To a normal 23-year-old subject in the luteal phase of the menstrual cycle was administered intravenously 10.2 μg of [4-14C]15α-hydroxyprogesterone containing 3.72  $\times$  106 dpm (specific activity 3.66  $\times$  105 dpm/μg). Urine was collected for 5 days and the conjugates present in each day's urine were first hydrolyzed by solvolysis and then with β-glucuronidase. A total of 4.8% of the injected radioactivity was excreted in the sulfate fraction and 57.6% in the glucosiduronate fraction of the urine. Since virtually all of the radioactivity was found in the first 4 day's urine, the extracts from this period were used for the isolation studies described.

The sulfate fraction (248 mg and  $1.78 \times 10^5$  dpm) was chromatographed on a 50-g silica gel column and four major pools of radioactivity, pools A-D, were obtained as described in the previous study. Metabolites eluted in pools B, C, and D have not as yet been identified. The glucosiduronate fraction (339 mg and  $2.14 \times 10^6$  dpm) was chromatographed on a 60-g silica gel column and six major pools of radioactivity, pools A-F, were obtained. Metabolites in pools A and E have not as yet been identified. Purification of the metabolites and their identification and proof of radiochemical purity were achieved by procedures identical with those described

for the metabolites isolated from pregnancy urine except that [ ${}^{3}$ H]acetic anhydride solution 2 was used for acetylation of the residues following the last chromatographic separation. Using these methods,  $3\beta$ ,15 $\alpha$ -dihydroxy-5 $\beta$ -pregnan-20-one was isolated from pool A of the sulfate fraction, and  $15\alpha$ -hydroxyprogesterone,  $3\alpha$ ,15 $\alpha$ -dihydroxy-5 $\alpha$ -pregnan-20-one,  $3\alpha$ ,15 $\alpha$ -dihydroxy-5 $\beta$ -pregnan-20-one, and  $5\alpha$ -pregnane-3 $\alpha$ ,-15 $\alpha$ ,20 $\beta$ -triol were isolated from pools B, C, D, and F, respectively, of the glucosiduronate fraction. The specific activities of the isolated metabolites are shown in Table IX.

#### Discussion

In order to study the metabolism of a tracer amount of labeled  $15\alpha$ -hydroxyprogesterone administered to a pregnant female, it was first necessary to obtain sizeable amounts of the urinary metabolites of this steroids, and to establish their identity. This was achieved in the first two studies in which five urinary metabolites of  $15\alpha$ -hydroxyprogesterone, namely,  $3\alpha$ ,  $15\alpha$ -dihydroxy- $5\alpha$ -pregnan-20-one,  $3\alpha$ ,  $15\alpha$ -dihydroxy- $5\beta$ -pregnan-20-one,  $3\beta$ ,  $15\alpha$ -dihydroxy- $5\beta$ -pregnan-20-one,  $5\alpha$ -pregnane- $3\alpha$ ,  $15\alpha$ ,  $20\beta$ -triol, and  $5\beta$ -pregnane- $3\alpha$ ,  $15\alpha$ ,  $20\beta$ -triol were isolated and identified.

The last two studies dealt with the metabolism of tracer doses of  $[4-1^4C]15\alpha$ -hydroxyprogesterone injected intravenously into a pregnant subject in the 38th week of gestation and in a nonpregnant subject in the luteal phase of the menstrual cycle. In both studies virtually all of the radioactivity excreted was found in the first 4 day's urine. Most of the radioactive conjugates were hydrolyzed with  $\beta$ -glucuronidase while only a small percentage of the excreted radioactive

metabolites was extracted following solvolysis. It is possible that some of the metabolites present in the glucosiduronate fraction were originally excreted as double conjugates with both sulfuric and glucuronic acids. As is evident from Table IX,  $3\beta$ ,  $15\alpha$ -dihydroxy- $5\beta$ -pregnan-20-one was isolated from the sulfate fraction in both studies, and  $15\alpha$ -hydroxyprogesterone,  $3\alpha$ ,  $15\alpha$ -dihydroxy- $5\alpha$ -pregnan-20-one,  $3\alpha$ ,  $15\alpha$ -dihydroxy- $5\beta$ -pregnan-20-one, and  $5\alpha$ -pregnane- $3\alpha$ ,  $15\alpha$ ,  $20\beta$ -triol were isolated from the glucosiduronate fraction of the urine.

The specific activities of the metabolites isolated from the urine of the pregnant subject (Table IX) were all from 25 to 45 times lower than the specific activity of the injected  $15\alpha$ hydroxyprogesterone demonstrating that all of these steroids are normal constituents of human late pregnancy urine. However, the specific activities of the metabolites isolated from the urine of the nonpregnant subject were almost identical with the specific activity of the injected  $15\alpha$ -hydroxyprogesterone, demonstrating that these steroids are not normally excreted in the urine of such subjects. In parallel studies we have found (Giannopoulos et al., 1969) that  $15\alpha$ -hydroxyprogesterone is not present in the urine during the luteal phase of the normal menstrual cycle, but is excreted during the second and third trimesters of pregnancy. It was also shown (Giannopoulos et al., 1969) that the conversion of progesterone to  $15\alpha$ -hydroxyprogesterone occurs during late pregnancy in the fetoplacental unit but not in the maternal circulation. Therefore, the results obtained in the pregnant and nonpregnant female are in agreement with these findings.

The specific activities of all the metabolites isolated from the urine of the pregnant subject were very similar with the exception of  $15\alpha$ -hydroxyprogesterone whose specific activity was significantly higher (Table IX). These results suggest that the metabolites of  $15\alpha$ -hydroxyprogesterone may in part be derived from precursors other than  $15\alpha$ -hydroxyprogesterone, such as  $15\alpha$ -hydroxypregnenolone which has not as yet been isolated. It must be borne in mind that  $[4-14C]15\alpha$ hydroxyprogesterone was injected into the maternal circulation and that during late pregnancy it is synthesized in the fetoplacental unit (Giannopoulos et al., 1969). It is therefore possible that  $15\alpha$ -hydroxyprogesterone is metabolized in the fetoplacental unit and that the metabolites cross the placenta and enter into the maternal compartment. If that is true, differences in the relative proportion of the fetal products and the metabolites of  $[4-14C]15\alpha$ -hydroxyprogesterone in the maternal compartment may explain the observed differences in the specific activities of the urinary metabolites.

Assuming that urinary  $15\alpha$ -hydroxyprogesterone is a unique metabolite of endogenous  $15\alpha$ -hydroxyprogesterone, the urinary production rate of this steroid in the maternal compartment may be calculated using the expression for a single compartment model (Pearlman *et al.*, 1954). Using this simple model the production rate was calculated to be 79  $\mu$ g/day. However, the calculated production rate may not reflect the true value and should be considered as a maximal value because of the possibility that other steroids, such as  $15\alpha$ -hydroxyprogenenolone (and its sulfate), may serve as important precursors of  $15\alpha$ -hydroxyprogesterone.

It is important to note that the major urinary metabolites of  $15\alpha$ -hydroxyprogesterone are the two dihydroxy ketones, namely  $3\alpha,15\alpha$ -dihydroxy- $5\alpha$ -pregnan-20-one and  $3\alpha,15\alpha$ -dihydroxy- $5\beta$ -pregnan-20-one, while very small amounts of

metabolites reduced at C-20 were isolated. In addition, a large proportion of the injected  $15\alpha$ -hydroxyprogesterone was excreted unchanged in the urine. These results suggest that the  $15\alpha$ -hydroxyl group inhibits the reduction of the 20-ketone and the  $\Delta^4$ -3-keto group of the steriod nucleus. As shown in Table IX, some of the reduced metabolites of  $15\alpha$ -hydroxyprogesterone are  $5\alpha$ -steroids. The overall ratio of  $5\alpha$ - to  $5\beta$ -reduced metabolites was found to be approximately 0.55. The generalization has been made that  $\Delta^4$ -3-keto  $C_{21}$  steroids are primarily reduced in vivo in man to the  $5\beta$  form (Dorfman, 1954). That the  $15\alpha$ -hydroxyl group in the  $C_{21}$  steroids influence the reduction in the A and B rings in favor of the  $5\alpha$ -metabolites was demonstrated by Ruse and Solomon (1966) who found that the  $5\alpha$ -urinary metabolites of labeled  $16\alpha$ -hydroxyprogesterone injected into a pregnant subject were present in amounts greater than would be expected for steroids of the  $C_{21}$  series. It seems, therefore, that the effect of the  $15\alpha$ hydroxyl group on the reduction of the  $\Delta^4$ -3-keto group of  $C_{21}$  steroids is similar to the effect of the  $16\alpha$ -hydroxyl constituent. Both hydroxyl groups favor an increased  $5\alpha/5\beta$  ratio of the reduced urinary metabolites but the  $15\alpha$ -hydroxy group has a greater inhibitory effect on the reduction of the  $\Delta^4$ -3keto. In addition, reduction of the 20 ketone of  $16\alpha$ -hydroxyprogesterone favors the  $20\alpha$ -alcohols since only  $20\alpha$ -triols of  $16\alpha$ -hydroxyprogesterone were isolated (Ruse and Solomon, 1966). On the other hand, the only  $C_{20}$ -reduced metabolites of  $15\alpha$ -hydroxyprogesterone isolated in these studies were  $5\alpha$ -pregnane- $3\alpha$ ,  $15\alpha$ ,  $20\beta$ -triol and  $5\beta$ -pregnane- $3\alpha$ ,  $15\alpha$ ,  $20\beta$ triol, indicating that the  $15\alpha$ -hydroxyl groups directs the reduction at C-20 toward the  $20\beta$ -alcohols.

It is of interest to note that  $15\alpha$  hydroxylation of neutral steroids is not confined to the human fetoplacental unit. Schneider (1965) found that liver slices of the American bullfrog (Rana catesbiana) was capable of converting deoxycorticosterone into a number of hydroxylated products, among which were  $15\alpha$ - and  $15\beta$ -hydroxydeoxycorticosterone. Recently, it has been found that the feces of conventional rats contain  $3\alpha$ ,  $15\alpha$ -dihydroxy- $15\alpha$ -pregnan-20-one as the predominant C-21 steroid (Gustafsson, 1968a,b) and that this metabolite is also present in the feces of the germ-free rat. In addition  $3\alpha$ ,  $11\beta$ ,  $15\alpha$ , 21-tetrahydroxy- $5\alpha$ -pregnan-20-one and  $3\alpha$ ,  $15\alpha$ , 21-trihydroxy- $5\alpha$ -pregnane-11,20-dione have been isolated as the major C-21 steroids from the feces of the germ-free rat (Gustafsson and Sjövall, 1968), and in trace amounts from the urine of the germ-free rat (Gustafsson, 1968a,b). Very recently  $15\alpha,20\beta$ -dihydroxypregn-4-en-3-one (10  $\mu$ g/beetle) was isolated from the secretion of the prothorax scent gland of the water beetle (Platambus macilatus) by Schildknecht et al. (1969). As a result of these findings it can be concluded that  $15\alpha$  hydroxylation of neutral steroids may be a common metabolic process in a large number of species.

#### Acknowledgments

We wish to acknowledge the assistance of Dr. Alan M. Duffield of the Chemistry Department, Stanford University, for the mass spectra and Drs. Yoko Kanazawa and Toshioki Mishida for the nuclear magnetic resonance spectra. Dr. Gunther Schilling helped with the optical rotations, and able technical assistance was provided by Mrs. Sheila Notkin and Miss Helen Blum. We would also like to thank Professor C. Djerassi for his interest and advice.

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# Isolation of $15\alpha$ -Hydroxyandrostenedione and $15\alpha$ -Hydroxytestosterone from Human Pregnancy Urine\*

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ABSTRACT: Five pools of urine were obtained from normal subjects in the third trimester of pregnancy. Purified [4- $^{14}$ C]- $^{15}\alpha$ -hydroxyandrostenedione or [4- $^{14}$ C]15 $\alpha$ -hydroxytestosterone was added to the urine as a recovery marker and the steroids were isolated and identified. The amount of 15 $\alpha$ -hydroxyandrostenedione excreted in the urine ranged from 4.4

to 5.2  $\mu$ g per day and that of  $15\alpha$ -hydroxytestosterone from 0.48 to 0.64  $\mu$ g per day, measured by the isotope derivative procedure using [³H]acetic anhydride. These findings demonstrate that  $15\alpha$ -hydroxyandrostenedione and  $15\alpha$ -hydroxytestosterone are normal excretory products in the third trimester of human pregnancy.

he formation of  $15\alpha$ -hydroxyestrogens from phenolic precursors in the pregnant female and by the newborn has been well documented (Schwers *et al.*, 1965; Hagen *et al.*, 1965; Gurpide *et al.*, 1966). When previable human fetuses were perfused with labeled androstenedione and testosterone, labeled  $15\alpha$ -hydroxyestradiol<sup>1</sup> was detected in the fetal liver (Mancuso *et al.*, 1968), thereby demonstrating that neutral  $C_{19}$ 

steroids can serve as precursors of  $15\alpha$ -hydroxyestrogens. In these latter studies  $C_{19}$   $15\alpha$ -hydroxysteroids could not be detected in the fetal tissues and the authors therefore concluded that  $15\alpha$  hydroxylation occurred after aromatization. With the finding that  $15\alpha$ -hydroxyandrostenedione can be aromatized by human placental tissue (Stern *et al.*, 1968) it appeared likely that  $15\alpha$ -hydroxyestrogens may be formed from  $15\alpha$ -hydroxy neutral precursors in an analogous manner to the

en-3 $\beta$ -yl sulfate; 15 $\alpha$ -hydroxyandrostenedione, 15 $\alpha$ -hydroxyandrost-4-ene-3,17-dione; 15 $\alpha$ -hydroxytestosterone, 15 $\alpha$ -hydroxyandrost-4-en-3-one; 15 $\alpha$ -hydroxyprogesterone, 15 $\alpha$ -hydroxypregn-4-ene-3,20-dione; deoxycorticosterone, 21-hydroxypregn-4-ene-3,20-dione; 15 $\alpha$ -hydroxyestradiol, estra-1,3,5(10)-triene-3,15 $\alpha$ ,17 $\beta$ -triol; 15 $\alpha$ -hydroxyestriol, estra-1,3,5(10)-triene-3,15 $\alpha$ ,16 $\alpha$ ,17 $\beta$ -tetrol; DDQ, 1,2-dichloro-5,6-dicyanobenzoquinone.

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 $<sup>^1</sup>$  The following trivial names and abbreviations are used: androstenedione, androst-4-ene-3,17-dione; testosterone,  $17\beta$ -hydroxyandrost-4-en-3-one; dehydroisoandrosterone sulfate, 17-oxoandrost-5-